This article was downloaded by:

On: 28 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



### Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713618290

# Novel Cations and Molecules from Phosphaalkynes, 1H-Phosphirenes and from Tetraphosphacubane

Kenneth K. Laali; Bernhard Geissler; Andreas Hoffmann; Wolfgang Fiedler; Thomas Mackewitz; Jürgen Simon; Sandro Hollenstein; Manfred Regitz

To cite this Article Laali, Kenneth K. , Geissler, Bernhard , Hoffmann, Andreas , Fiedler, Wolfgang , Mackewitz, Thomas , Simon, Jürgen , Hollenstein, Sandro and Regitz, Manfred (1999) 'Novel Cations and Molecules from Phosphaalkynes, 1H-Phosphirenes and from Tetraphosphacubane', Phosphorus, Sulfur, and Silicon and the Related Elements, 144: 1, 281 - 284 - 284 - 285 - 286 - 286 - 286 - 286 - 287 - 287 - 287 - 287 - 287 - 287 - 287 - 287 - 287 - 287 - 288 - 287 - 288 - 288 - 288 - 289 - 289 - 289 - 289 - 289 - 289 - 280 - 280 - 280 - 280 - 280 - 280 - 280 - 280 - 281 - 28

To link to this Article: DOI: 10.1080/10426509908546236 URL: http://dx.doi.org/10.1080/10426509908546236

#### PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

## Novel Cations and Molecules from Phosphaalkynes, 1H-Phosphirenes and from Tetraphosphacubane

KENNETH K. LAALI<sup>a</sup>, BERNHARD GEISSLER<sup>ab</sup>, ANDREAS HOFFMANN<sup>ab</sup>, WOLFGANG FIEDLER<sup>ab</sup>, THOMAS MACKEWITZ<sup>ab</sup>, JÜRGEN SIMON<sup>b</sup>, SANDRO HOLLENSTEIN<sup>a</sup> and MANFRED REGITZ<sup>b</sup>

<sup>a</sup>Department of Chemistry, Kent State University, Kent, OH 44242, USA and <sup>b</sup>Department of Chemistry, University of Kaiserslautern, D-67663 Kaiserslautern, Germany

Generation and trapping (counter ion) of the phosphavinyl cation (the P-analog of vinyl cation) by low temperature protonation of kinetically stable phosphaalkynes in super acid media led to the formation of the corresponding phosphaalkenes without direct observation of a long-lived cation. Reactions of RCP (R = tBu, 1-Ad) with "PhSeF" and with PhSeCI resulted in bis-phenylselenenylation and chlorophenylselenenylation respectively, forming novel Se-containing phosphaalkenes. Ionization of 1-H-triflato-phosphirene with B(OTf) 3 /SO2 led to the direct observation of a persistent phosphirenylium cation. The (CO)<sub>5</sub> W-complexed phosphirenylium cation was subsequently generated from the (CO)<sub>5</sub> W-complexed triflate and trifluoroacetate derivatives. Progress in P-functionalization of tetraphosphacubane by reactions with potent electrophiles (i.e. MeOTf, EtOTf PhCH<sub>2</sub>OTf', TMSCH<sub>2</sub>OTf), alkynyl-iodonium triflates and with protic superacids are briefly summarized.

Keywords: phosphaalkynes; 1H-phosphirenes; tetraphosphacubane; electrophilic chemistry; superacids; stable phosphorus cations; Se-containing phosphaalkenes

Thanks to the highly versatile chemistry of phosphaalkynes, they have developed into an important class of low-coordination organophosphorus compounds which can function as building blocks for the rapid assembly of numerous new molecules and cations. Several review articles and book chapters have summarized the progress which has been achieved in this field over the past two decades. <sup>1-7</sup> Utilizing the synthetic methods developed by Regitz and coworkers, beginning with two kinetically stable phosphaacetylenes 1-2 which are more conveniently accessible, various cyclooligomers (such as 3 and 4) can be synthesized chemoselectively via cycloaddition reactions. Cycloaddition also provides access to 5 as a key derivative for phosphirene chemistry (Fig 1). In relation to our work in carbocation and onium ion chemistry, we have been involved in electrophilic chemistry of the above compounds and their derivatives as a means to generate interesing phosphorus cations and molecules. <sup>8-14</sup>

Fig 1 +CEP 
$$\frac{P}{1}$$
  $\frac{P}{1}$   $\frac{P$ 

A persistent phosphavinyl cation has not yet been directly observed; low temperature protonation of 1-2 produced the corresponding phosphaalkenes by initial C-protonation and counter ion quenching to give phosphaalkenes by anti addition (Fig. 2). High level theoretical calculations showed that C-protonation is favored over P-protonation and at higher levels a cyclic cation was found to be the minimum. 12

Fig 2
$$CEP \xrightarrow{FSO_2H/SO_2CIF} \left[ \begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \end{array} \right] \xrightarrow{C=P_0} \begin{array}{c} \\ \\ \\ \\ \\ \\ \end{array} \xrightarrow{C=P_1} \begin{array}{c} \\ \\ \\ \\ \\ \end{array} \xrightarrow{CSO_2F}$$

In experiments designed to produce phosphaalkenes by fluoroselenenylation of 1-2 with "PhSeF" via (PhSe)<sub>2</sub>/XeF<sub>2</sub>, double phenyl selenenylation was observed, whereas with PhSeCl, chloroselenenylation was the outcome but with "PhSe<sup>+</sup>" attacking at P (Fig 3). 13

Fig 3 P(Fig 3): 13
$$(PhSe)_2 + XeF_2 \frac{CH_2CI_2}{\cdot 20^9 C \cdot (\cdot Xe)} 2 \left[ PhSeF \right] \frac{+ C \equiv P(2 \text{ equiv})}{\cdot 20^9 C \cdot 20 \text{ min}} \bigvee_{PhSe} C = P-SePh$$

$$2 \left[ PhSeF \right] + \bigcup_{C \equiv P} \frac{\cdot 20^9 C \cdot > 11}{\cdot 20^9 C \cdot 20 \text{ min}} \bigvee_{PhSe} C = P-SePh$$

$$PhSeCI \frac{R-C \equiv P}{C} \stackrel{R}{\leftarrow} C = P-SePh$$

In the field of phosphirene chemistry, ionization of the triflate derivative with B(OTf)<sub>3</sub> in SO<sub>2</sub> produced a persistent phosphirenylium cation, deshielded by 313 ppm relative to the precursor (Fig 4)<sup>11</sup>; multinuclear NMR and theory suggest aromatic character. Solvation with SO<sub>2</sub> is a key factor in lowering the energy barrier to ionization. The topic has been examined in detail computationally. The W-complexed analogue has also been generated and studied by multinuclear NMR (Fig 4). In the solution of the triflate derivative with B(OTf)<sub>3</sub> in SO<sub>2</sub> produced a produced and studied by multinuclear NMR (Fig 4).

Fig 4

$$P = OTI = \frac{B(OTI)_3}{SO_2 : -78^{\circ}C}$$
 $P = OTI = \frac{B(OTI)_3}{SO_2 : -78^{\circ}C}$ 
 $P = OTI = \frac{B(OTI)_3}{SO_2 : -78^{\circ}C}$ 

Focusing on the electrophilic chemistry of 3, decreased basicity/nucleophilicity at P is due to n P ->  $\sigma$  PC delocalization, thus showing a novel P/C bonding state with net positive charge at P and net negative charge at the cage carbons.<sup>4</sup>

Monoalkylation at P was achieved using potent alkylating agents such as MeOTf, EtOTf, "PhCH<sub>2</sub>OTf' and TMSCH<sub>2</sub>OTf to furnish stable mono-phosphonium salts, mono- and diprotonation were effected in superacids at low temperature, and monoalkynylation was achieved with electron-deficient alkynes to give the P-alkynylated salts (Fig 5):<sup>8,9</sup>

#### References

- [1] M. Regitz Chem. Rev., (1990), 90, 191.
- [2] M. Regitz in "Heteroatom Chemistry" Block, E., Ed, VCH, New York, (1990), chap. 17.
- [3] M. Regitz,; A. Hoffmann,; U. Bergstrasse in "Modern Acetylene Chemistry"; P. J. Stang, and F. Diederich, Eds. VCH, New York, (1995).
- [4] M. Regitz in "Organic Synthesis via Organometallics"; Proceedings of the Fourth Symposium in Aachen, July 1992; D. Enders; H. J. Gais; W. Kein, Eds., (1992) and related references cited therein.
- [5] M. Regitz Bull. Soc. Chim. Belg. (1992), 101, 361.
- [6] H. Memmesheimer; M. Regitz Reviews on Heteroatom Chemistry, (1994), 10, 61.

- [7] F. Mathey, M. Regitz in "Comprehensive Heterocyclic Chemistry II"; A. R. Katritzky; C. W. Rees; E. F. Scriven, Eds., Pergamon, (1998), Vol 1A, p.277.
- [8] K. K. Laali; M. Regitz; M. Birkel; P. J. Stang.; C. M. Crittell J. Org. Chem., (1993), 58, 4105.
- [9] K. K. Laali; B. Geissler; M. Regitz; J. J. Houser J. Org. Chem., (1995), 60, 47.
- [10] K. K. Laali; B. Geissler; M. Regitz J. Org. Chem., (1995), 60, 3149.
- [11] K. K. Laali; B. Geissler; O. Wagner; J. Hoffmann; R. Armbrust; W. Eisfeld; M. Regitz J. Am. Chem. Soc., (1994), 116, 9407.
- [12] K. K. Laali; B. Geissler; M. Regitz; J. J. Houser J. Org. Chem., (1995), 60, 6362.
- [13] K. K. Laali; W. Fielder; M. Regitz Chem. Commun. (1997), 1641.
- [14] K. K. Laali; C. Wesdemiotis; M. Polce; S. Beranova J. Chem. Soc. Perkin, Trans. 2, (1995), 1191.
- [15] W. Eisfeld; M. Regitz J. Org. Chem. (1996), 63, 2814.
- [16] J. Simon; A. Hoffmann; M. Regitz; K. K. Laali manuscript in preparation.